

# Practical Use of Corrected Fluorescence Excitation and Emission Spectra of Fluorescent Proteins in Förster Resonance Energy Transfer (FRET) Studies

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Corrected fluorescence excitation and emission spectra have been obtained from several enhanced variants of the green fluorescent protein (EGFP) isolated from the jellyfish *Aequorea victoria*, blue fluorescence protein (EBFP), cyan fluorescent protein (ECFP), EGFP and yellow fluorescent protein (EYFP–citricine) and from the red fluorescent protein (DsRed) isolated from the coral species *Discosoma*. The spectra are stored in a database. This report describes how the spectra can be used as templates to derive the critical transfer distance for any pair of fluorescent proteins.

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**KEY WORDS:** Fluorescent proteins; FRET; GFP; DsRed; resonance energy transfer.

## INTRODUCTION

The green fluorescent protein (GFP) from the jellyfish *Aequorea victoria* together with its differently colored mutants has received widespread use as a natural, brightly fluorescent marker in cell biology [1–3]. Pairs of fluorescent proteins are widely used as donor–acceptor or Förster resonance energy transfer (FRET) pairs and are as such fused to the protein(s) of interest and genetically encoded in cells. FRET is a very powerful method for obtaining distance information on macromolecular complexes. It is based on the phenomenon that excited-state energy from a donor to an acceptor molecule is transferred nonradiatively through space [4]. Note that donor and acceptor molecular dipoles are very weakly inter-

acting, which suggests that they can be neither nearest neighbours nor positioned at van der Waals distance from each other. FRET has been developed during the last decades as a spectroscopic ruler in the range of 2–7 nm to determine donor–acceptor distances in biological macromolecules (reviewed in, among others, references [5–9]). For a review of the principles of FRET applied to cell biology and microscopy, see references [10–14].

We collected corrected fluorescence excitation and emission spectra of several enhanced variants of green fluorescent protein (EGFP): blue fluorescence protein (EBFP), cyan fluorescent protein (ECFP), green fluorescent protein (EGFP), yellow fluorescent protein (EYFP–citricine), all from *Aequorea victoria*, and the red fluorescent protein from the coral species *Discosoma* (DsRed). From these spectra the parameters required for quantization of resonance energy transfer (overlap integral, critical transfer or Förster distance) can be easily recovered. We have made use of these spectra in various undergraduate and graduate courses to demonstrate and exercise the calculation of the spectral overlap integral and critical transfer distance. In this report, we give an account of this method.

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## MATERIALS AND METHODS

The isolation of these proteins was performed as described in ref. [15]. The steady-state fluorescence spectra were obtained with a Spex-Fluorolog 3.2.2 (Horiba Jobin Yvon, Longjumeau, France) spectrofluorometer at room temperature (295 K). The concentration of all fluorescent proteins was in the range 50–100 nM and a quartz cuvette of  $1.0 \times 0.4$  cm contained the protein solution. For all FPs except EBFP, the slits in the double-grating excitation and emission monochromators were 1 nm. For EBFP, these slits were 2 nm. Both excitation and emission spectra were corrected by a procedure supplied by the manufacturer and normalized to unity using the maximum values. The spectra were acquired at wavelength steps of 1 nm.

The rate of resonance energy transfer  $k_T$  is given by the Förster rate equation

$$k_T = (1/\tau_D^0) (R_0/R)^6 \quad (1)$$

where  $\tau_D^0$  is the donor fluorescence lifetime in the absence of acceptor,  $R$  is the actual donor-acceptor distance, and  $R_0$  is the critical transfer distance (or Förster distance) at which the rate of transfer is equal to the fluorescence decay rate:

$$R_0^6 = 8.785 \times 10^{-5} \kappa^2 Q_D^0 J/n^4 \quad (2)$$

where  $\kappa^2$  is the orientation factor between donor and acceptor molecules,  $Q_D^0$  is the quantum yield of donor fluorescence in the absence of acceptor, and  $n$  is the index of refraction, usually taken as 1.4.  $J$  is the overlap integral between the fluorescence spectrum of the donor and the molar absorption spectrum of the acceptor:

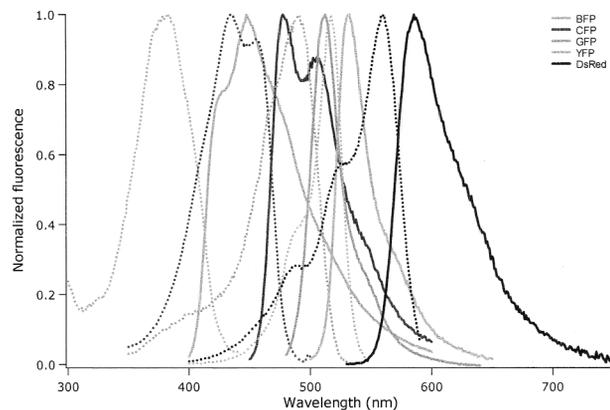
$$J = \int F_D(\lambda) \epsilon_A(\lambda) \lambda^4 d\lambda \quad (3)$$

where  $F_D$  is the peak-normalized fluorescence spectrum of the donor and  $\epsilon_A$  is the absorption spectrum of the acceptor. The critical transfer distance  $R_0$  is in units of angstroms, whereas  $\lambda$  is in nanometers and  $\epsilon_A$  in  $\text{M}^{-1}\text{cm}^{-1}$ .

Equations (2) and (3) were programmed in Igor Pro (Wavemetrics Inc., Lake Oswego, OR).

## RESULTS AND DISCUSSION

Corrected excitation and emission spectra of all FPs are shown in Fig. 1. From these spectra one can easily evaluate the critical transfer distance  $R_0$  and the spectral overlap integral  $J$  provided that the quantum yield of the donor fluorescence  $Q_D^0$  and the molar extinction coefficient of the acceptor at the wavelength of maximum light absorption  $\epsilon_A(\lambda_{max})$  are known. These



**Fig. 1.** Corrected excitation (dashed lines) and emission (solid lines) spectra of (from left to right) EBFP, ECFP, EGFP, EYFP, and DsRed.

data are known from the literature [1,16] and are summarized in Table I. A typical protocol for evaluation of transfer parameters in Igor Pro is given in the procedure section for the donor-acceptor pair ECFP-EYFP. After the data files are read in (emission spectrum of ECFP, excitation spectrum of EYFP, and the respective wavelengths), the emission spectrum of ECFP is divided by its integral, so that the integrated spectrum  $[\int F_D(\lambda) d\lambda]$  is unity. Then, the excitation spectrum of EYFP is multiplied by its extinction coefficient (Table I) to yield the molar absorption spectrum of the acceptor. This is followed by the calculation of the spectral overlap integral  $J$ , in which integration is performed using a trapezoidal algorithm. Finally, the critical transfer distance  $R_0$  is calculated according to Eq. (2) using the quantum yield  $Q_D^0 = 0.39$  of CFP, the calculated  $J$ , the refractive  $n = 1.4$ , and the orientation factor  $\kappa^2 = 1$  (antiparallel transition dipoles of donor and acceptor).

**Table I.** Useful Numbers for Calculation of Spectral Overlap Integral and Critical Transfer Distance Between Fluorescent Proteins<sup>a</sup>

FP	Donor quantum yield (-)	Acceptor extinction coefficient ( $\text{M}^{-1}\text{cm}^{-1}$ )
EBFP (F64L, Y66H, Y145F)	0.24	22,000
ECFP (F64L, S65T, Y66W, N146I, M153T, V163A)	0.39	32,500
EGFP (F64L, S65T)	0.64	56,000
EYFP-Citrine (S65G, V68, Q69K, S72A, T203Y)	0.61	62,000
DsRed	0.70	75,000

<sup>a</sup> The data for EBFP, ECFP, EGFP, and EYFP are from reference [1], the data for DsRed are from reference [16].

**Table II.** Numerical Values of Spectral Overlap Integral  $J$  and Critical Transfer Distance  $R_0$  for Various Donor–Acceptor Pairs of Fluorescent Proteins

Donor–acceptor pair	$J$ ( $M^{-1} cm^3$ )	$R_0$ ( $\kappa^2 = 1$ ) (Å)	$R_0$ ( $\kappa^2 = 4$ ) (Å)
EBFP → EGFP	$1.24614 \times 10^{15}$	44	55
ECFP → EGFP	$1.60525 \times 10^{15}$	49	62
ECFP → EYFP	$1.55479 \times 10^{15}$	49	62
EGFP → EGFP	$1.02816 \times 10^{15}$	49	62
EGFP → EYFP	$2.30938 \times 10^{15}$	57	71
EGFP → DsRed	$2.88576 \times 10^{15}$	59	74
EYFP → EYFP	$9.76207 \times 10^{14}$	49	62
EYFP → DsRed	$3.92001 \times 10^{15}$	62	78

In this way, all parameters for various donor–acceptor pairs have been evaluated from the available spectra and data; these values have been collected in Table II. We have also listed in Table II the  $R_0$  values when the maximal value of the orientation factor is reached, namely for parallel transition dipoles ( $\kappa^2 = 4$ ). Some interesting conclusions can be drawn from the data in Table II. The more red-shifted the absorption spectra, the higher are the molar extinction coefficients (Table I). In addition, the more red-shifted the fluorescence spectrum, the higher is the quantum yield (Table I). Therefore, the overlap integrals exhibit a steady increase when the red-shifted fluorescent proteins act as acceptor. This increase is reflected as well in a larger critical distance  $R_0$ . Homo-transfer between different EGFPs (or between different EYFPs) can occur quite readily, as judged from the appreciable  $J$  and  $R_0$  values. Homo-transfer of GFP fusion proteins in cells has been described recently via fluorescence anisotropy decay measurements [17].

**PROCEDURE**

The annotated procedure for Igor Pro was used to calculate the overlap integral and critical transfer distance from experimental fluorescence data.

General text load from  
 “cfpem.dat” *loading in experimental data*  
 Data length: 151,  
 waves: cfpem *(151 data points per file)*  
 General text load from  
 “wlcfp.dat”  
 Data length: 151,  
 waves: wlcfp  
 General text load from  
 “yfpex.dat”

Data length: 151, waves:  
 yfpex  
 General text load from  
 “wlyfpex.dat”  
 Data length: 151, waves:  
 wlyfpex  
 •display cfpem vs  
 wlcfp *make graph of 2 spectra*  
 •append yfpex vs  
 wlyfpex  
 •duplicate cfpem  
 cfpintegr *duplicate in order not to  
 overwrite original  
 data*  
 •duplicate wlcfp  
 wlcfpmod  
 •duplicate yfpex  
 yfpexnorm  
 •duplicate wlyfpex  
 wlyfpexmod  
 •duplicate cfpem  
 cfpemnorm  
 •integrate/T cfpintegr *integration of emission  
 spectrum*  
 •edit cfpintegr *get value of integral*  
 •cfpemnorm=cfpemnorm/  
 68.1649 *division by value of  
 integral*  
 •yfpexnorm=yfpexnorm\*  
 62000 *multiplication with  
 $\epsilon_{max}$*   
 •edit wlcfpmod,cfpemnorm,  
 wlyfpexmod,yfpexnorm  
 •DeletePoints 0,50,  
 yfpexnorm, wlyfpexmod *shift of absorption  
 spectrum such  
 that spectra start at  
 same  $\lambda$*   
 •duplicate wlyfpexmod  
 overlap  
 •overlap=cfpemnorm\*  
 yfpexnorm\*  
 wlyfpexmod^4 *overlap integral J  
 (eq. 3)*  
 •duplicate overlap overla-  
 pint  
 •integrate/T overlapint *integration of J*  
 •edit overlapint *get value of J*  
 •make/d rzero  
 •rzero=8.785e-5\*1\*0.39\*  
 1.55479e15/(1.4^4) *calculation of  $R_0$   
 [Eq. (2)]*  
 •rzero=rzero^0.1666666  
 •display rzero *get value of  $R_0$*

## CONCLUSION

When the corrected excitation and emission spectra of different fluorescent proteins are used, it is rather straightforward to recover the correct energy transfer parameters. All spectra in digital form are available upon request.

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